

Controlling dye delivery in electronic imaging via dye-polymer interactions

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| Agenda

1. Introduction to D2T2
2. Dye-polymer interactions in the donor layer
3. Dye-polymer interactions in the acceptor layer
4. Dye-dye interactions...
5. Conclusions

| Dye Diffusion Thermal Transfer (D2T2) Printing

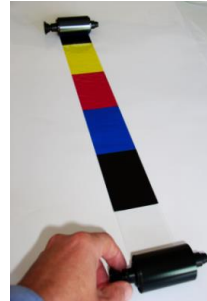
- Photo realistic images - continuous tone printing process
- High resolution, 20 OD levels per colour – 8 million colours
- High print speeds – 6 x 4 inch photo printed in 5 seconds

- Major Applications
 - ✓ Digital photo finishing – Photobooths, Kiosks
 - ✓ Novelty photo – Events and Rides
 - ✓ ID Card personalisation – National ID Cards, Driving Licences, Credit Cards

| ITW Imagedata - D2T2 Product Assembly for ID Cards

YMC Dye panels

2 dyes + PVB/EC binder
Diffusion of dye into card
release from card
control of delivery rate of dye
have to compensate for back diffusion



•K Panel - IR readable text & barcodes
Carbon black + resins

Overlay - image protection
Polyester resin

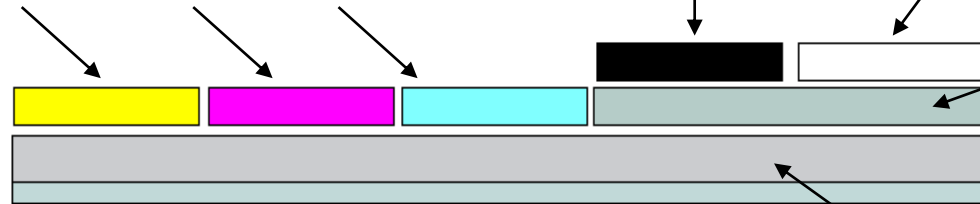
Sub coat - release of K
& Overlay
UV Cured acrylic

Back coat

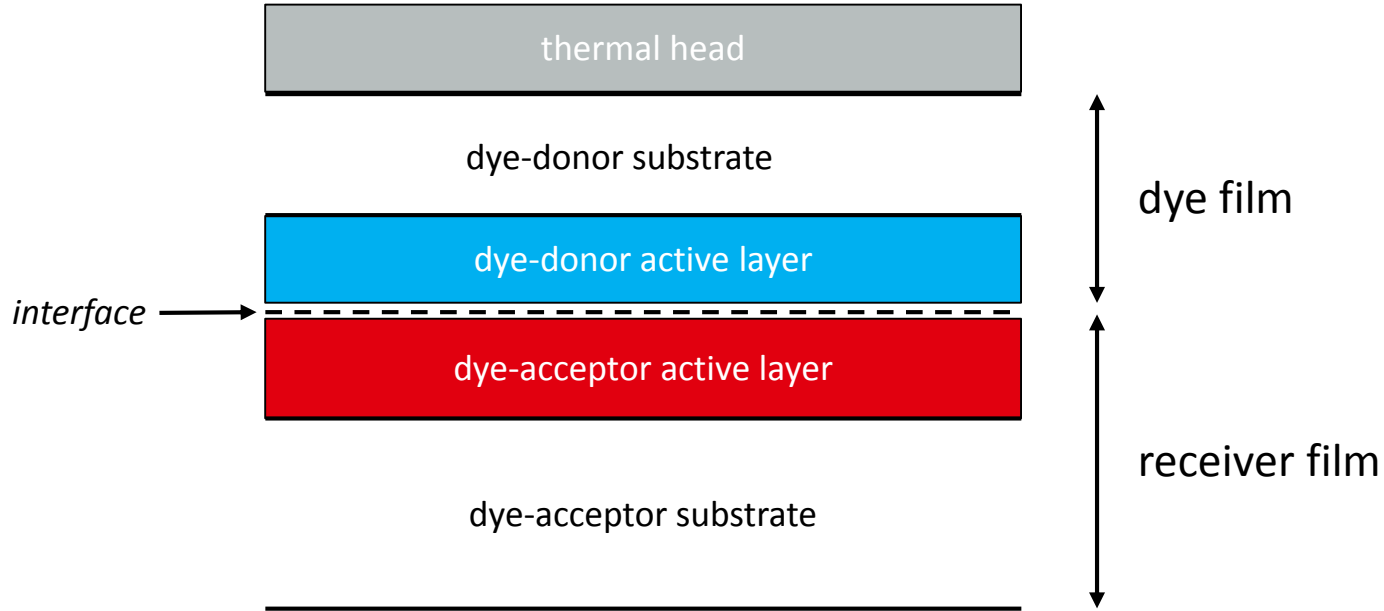
UV cured acrylic
solid lubricant
lubricates passage of thermal head over media
must not corrode ceramic on the head

PET substrate

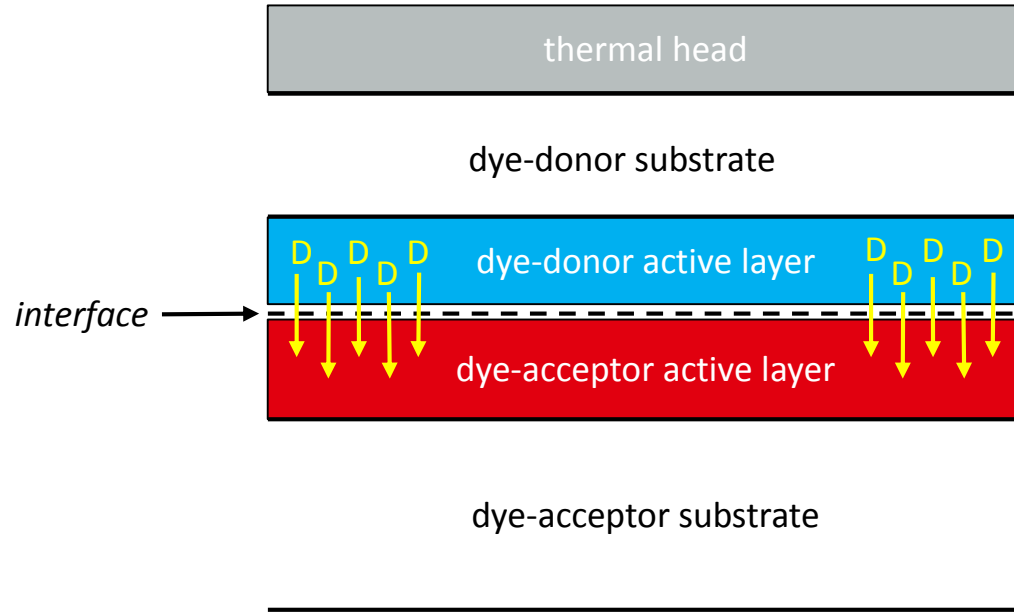
4.5 / 6um thick
adhesion pre-treated
pre-tensilised



| Dyes transfer via diffusion from donor to acceptor layers



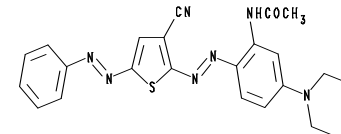
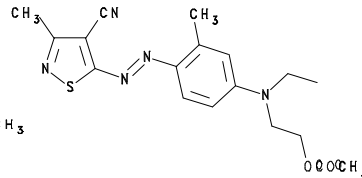
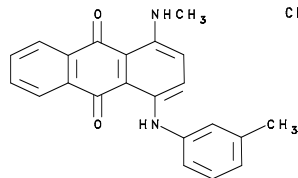
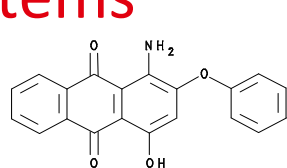
| Dyes transfer via diffusion from donor to acceptor layers



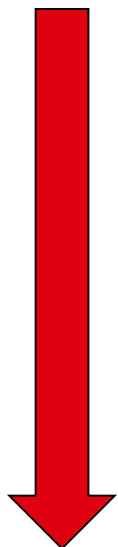
- dyes transfer by diffusion not sublimation
- temperature cycle 150 – 300 °C for milliseconds
- constant print time = fixed temperature cycle

Important to control dye transfer for colour density and colour balance

| Dye-donor systems

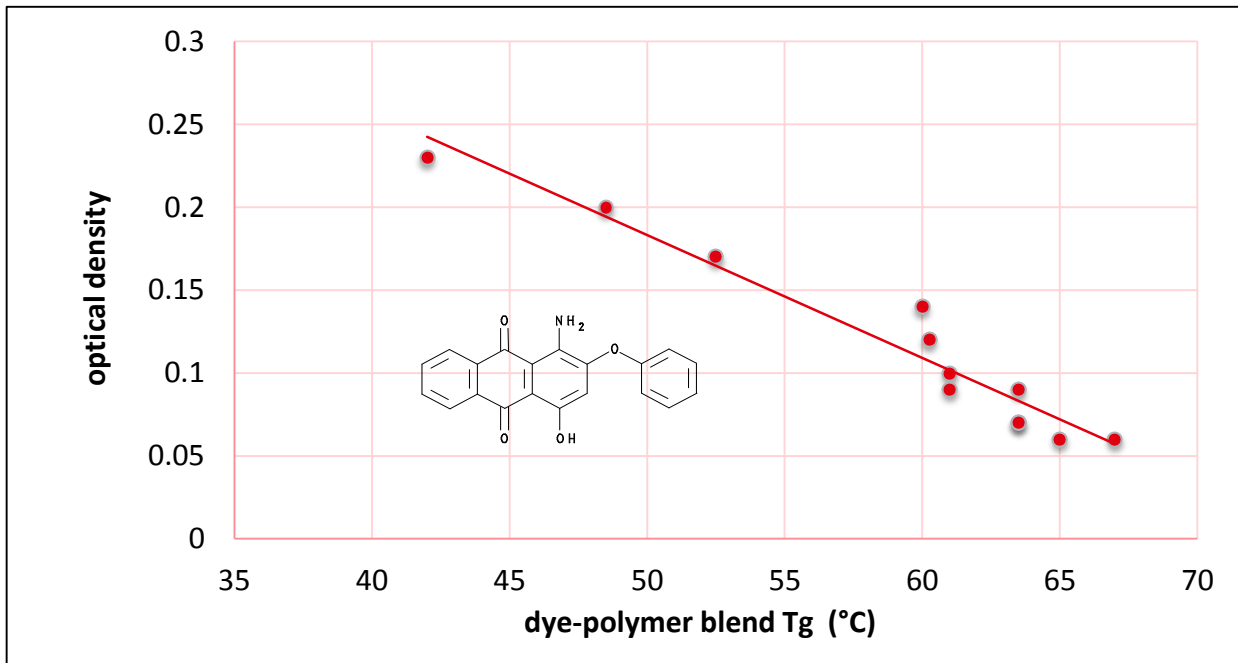


polymer	Tg (°C)	d1-donor Tg (°C)	d2-donor Tg (°C)	d3-donor Tg (°C)	d4-donor Tg (°C)
CAB	93.0	42.0	-----	-----	-----
EC	93.0	48.5	38.0	26.5	85.0
PS	104.5	52.5	57.0	34.0	95.0
PVC	82.5	60.0	-----	41.5	-----
PC	100.0	61.5	-----	45.5	-----
PEST	100.0	61.0	-----	45.0	-----
PHEN	78.5	61.0	-----	46.0	-----
CPVC	99.0	63.5	65.5	53.0	97.0
PVB	85.0	-----	61.0	39.0	89.0
SAN	103.0	63.5	64.0	54.0	97.0
PVAA	96.0	65.0	72.0	52.0	98.0
PVF	86.5	67.0	65.5	47.5	91.0
PPHS	145.0	-----	80.0	-----	120.0



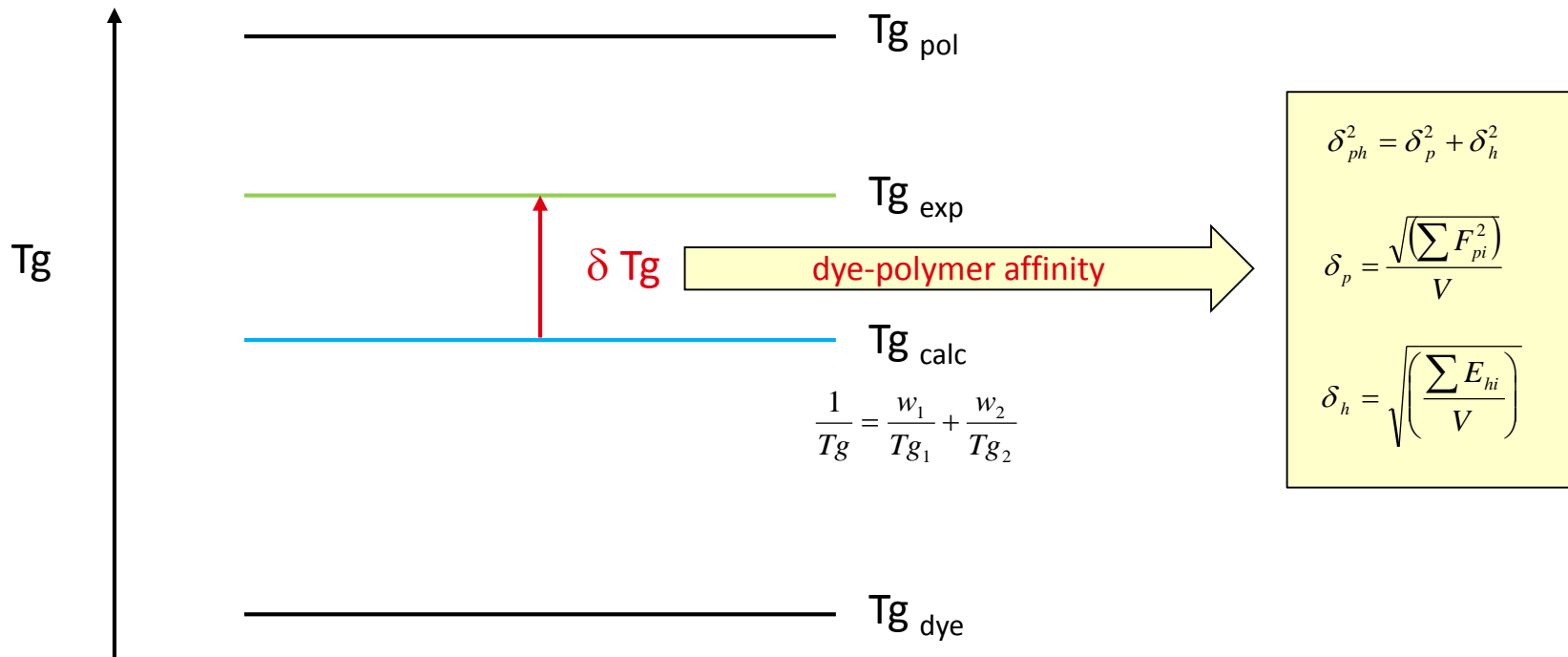
increasing
polarity

| Dye transfer controlled by Tg of dye-polymer blend

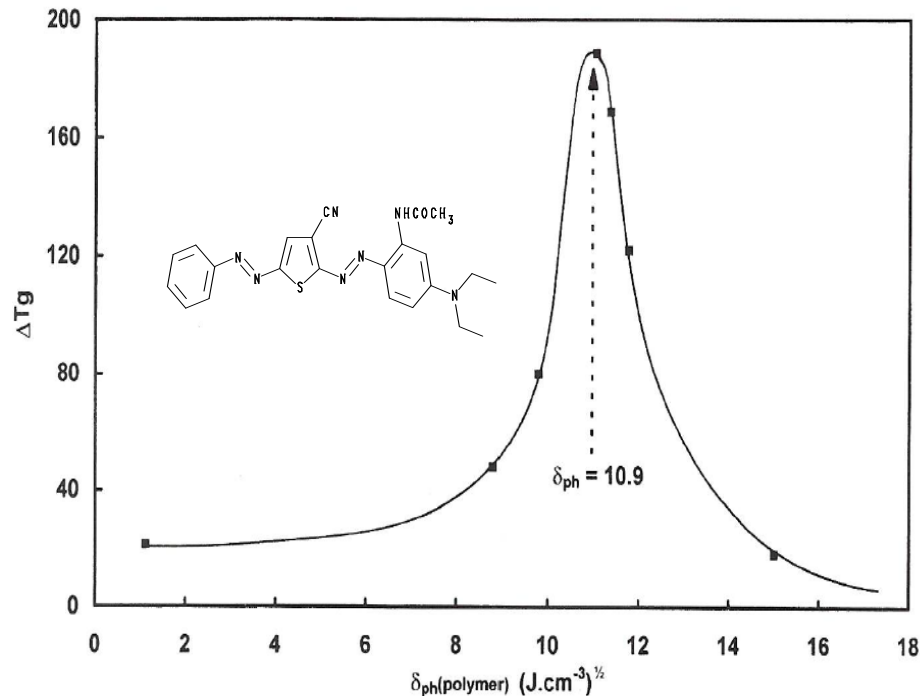
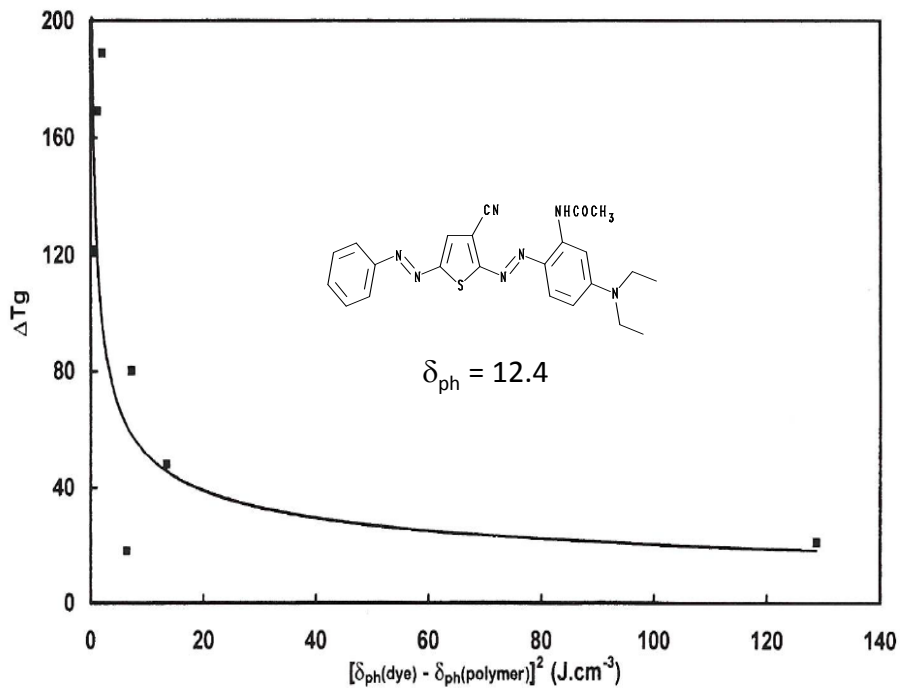


Similar behaviour observed for other dyes

| Model of factors influencing Tg

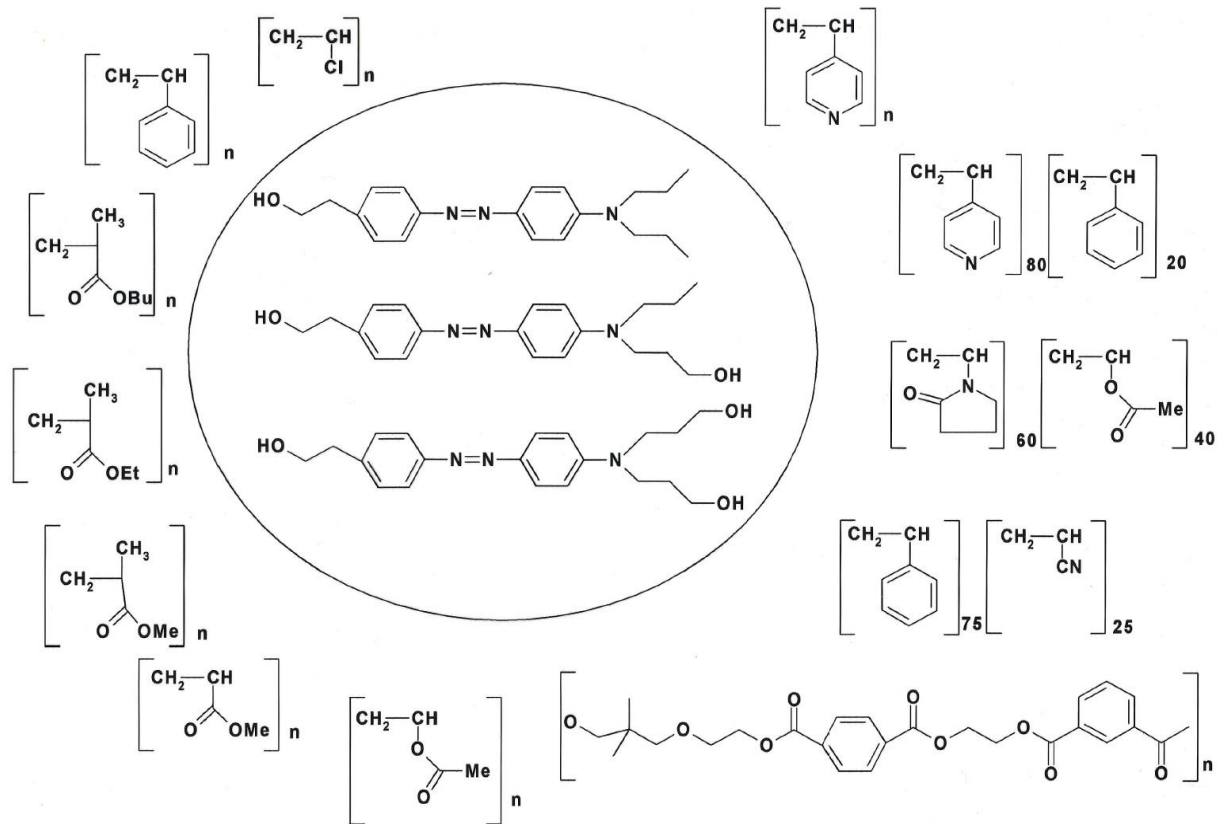


Higher Tg from stronger dye-polymer interactions



Similar behaviour observed for other dyes

| Dye-acceptor systems



Development of Solution-Diffusion Model

Permeability = Diffusivity x Solubility

$$P = D.S$$

diffusion coefficient relates to free volume

$$D = RTA_d \cdot \exp\left(\frac{-B_d}{f}\right)$$

solubility coefficient relates to heat of solution

$$S = S_0 \cdot \exp\left(\frac{-\Delta H}{RT}\right)$$

free volume depends on difference between transport temperature and T_g

$$f = f_g + \alpha(T - T_g)$$

$$\Delta H = \Delta H_{\text{physical}} + \Delta H_{\text{specific}}$$

endothermic:
solubility parameter difference

$$\Delta H_{\text{physical}} = \phi_d \phi_p (\delta_d - \delta_p)^2$$

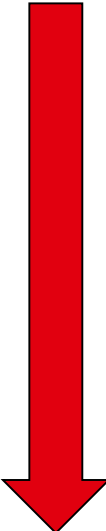
exothermic:
specific interactions

$$\Delta H_{\text{specific}} = \text{function}(v_{\text{inter}} - v_{\text{free}})$$

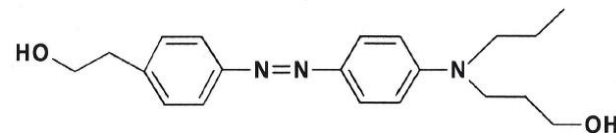
$$\ln P = a_4 + b_4 \left(\frac{1}{Tg}\right) - c_4 (\delta_d - \delta_p)^2 - d_4 (v_{\text{inter}} - v_{\text{free}})$$

| Bihydroxyl dye transfer to different acceptor polymers

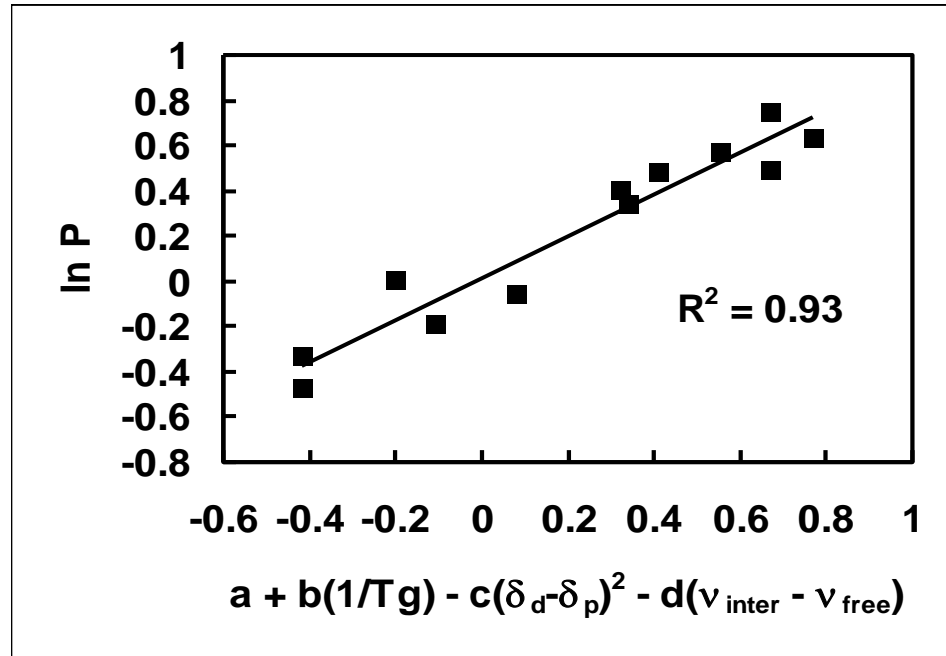
polymer	functional group	Tg (°C)	O-H - X (cm ⁻¹)
PVC	-Cl	88.0	3602
PS	-phenyl	110.8	3583
PBMA	-COO-	71.5	3544
PES1	-COO-	47.0	3547
PES2	-COO-	69.0	3547
PMMA	-COO-	122.0	3541
PEMA	-COO-	80.0	3541
PMA	-COO-	19.5	3541
PVAc	-COO-	41.7	3520
P(S-co-AN)	-CN, -phenyl	110.3	3518
P(VPy-co-VAc)	-CON-, -COO-	61.6	3368
P(VP-co-S)	-N=, phenyl	97.0	3281
PVP	-N=	90.0	3279



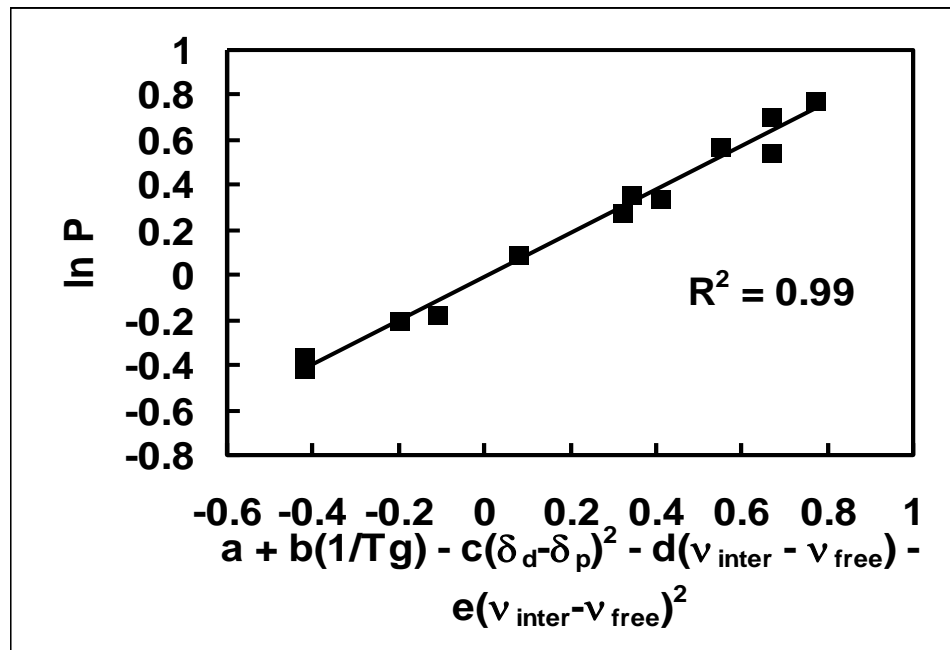
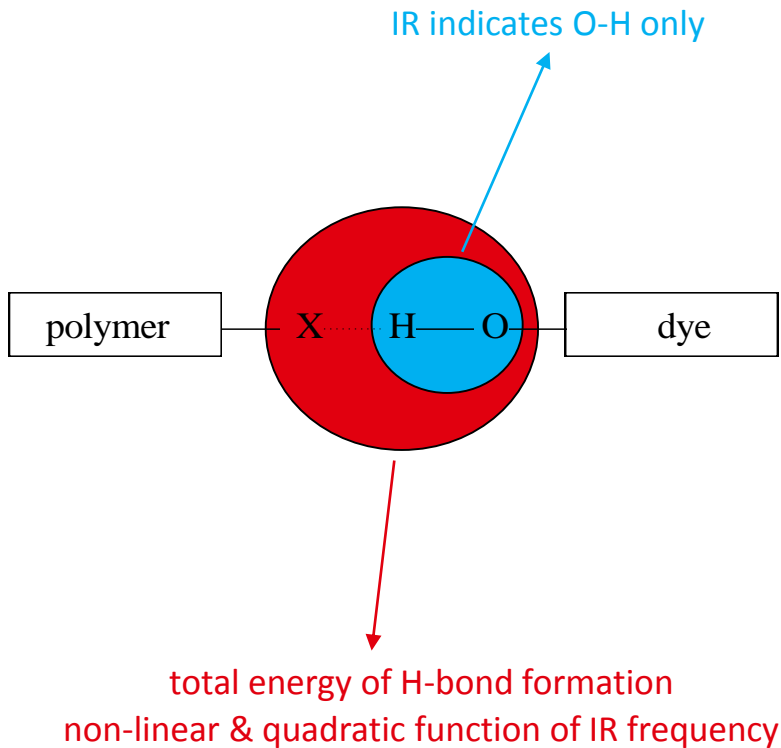
increasing
specific
interaction



| Bihydroxyl dye transfer to different acceptor polymers

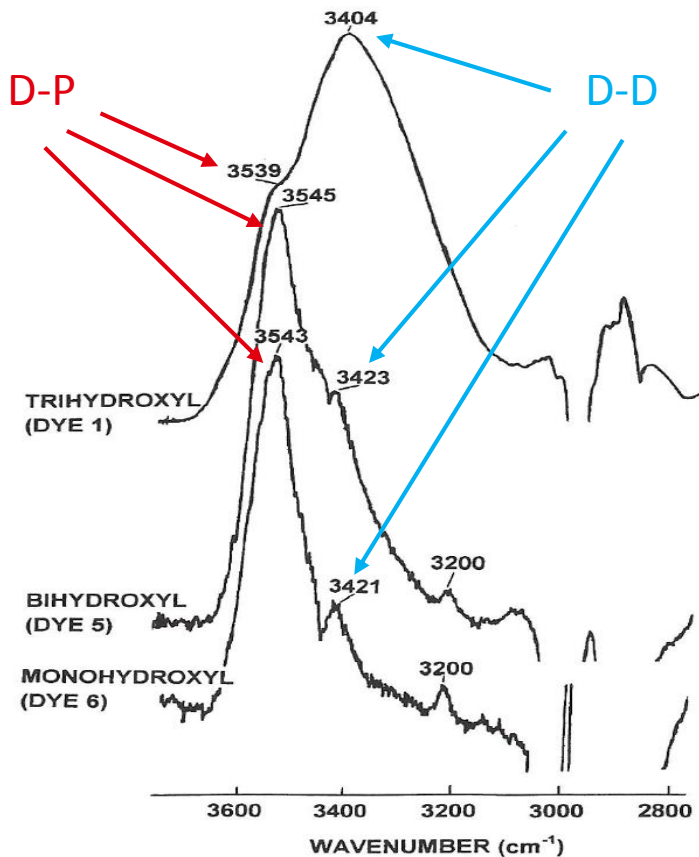


| Bihydroxyl dye transfer to different acceptor polymers



| Dye-dye interactions.....dye-polymer interactions

various dyes: fixed polyester



D-P interaction strength

- similar for different dyes

D-D interaction strength

- D-D always stronger than D-P
- hydroxyl $F3 > (F2 \sim F1)$

D-D: D-P ratio

- hydroxyl $F3 \gg F2 > F1$

| Dye-dye interactions.....dye-polymer interactions

fixed dye: various polymers

D-P interaction strength

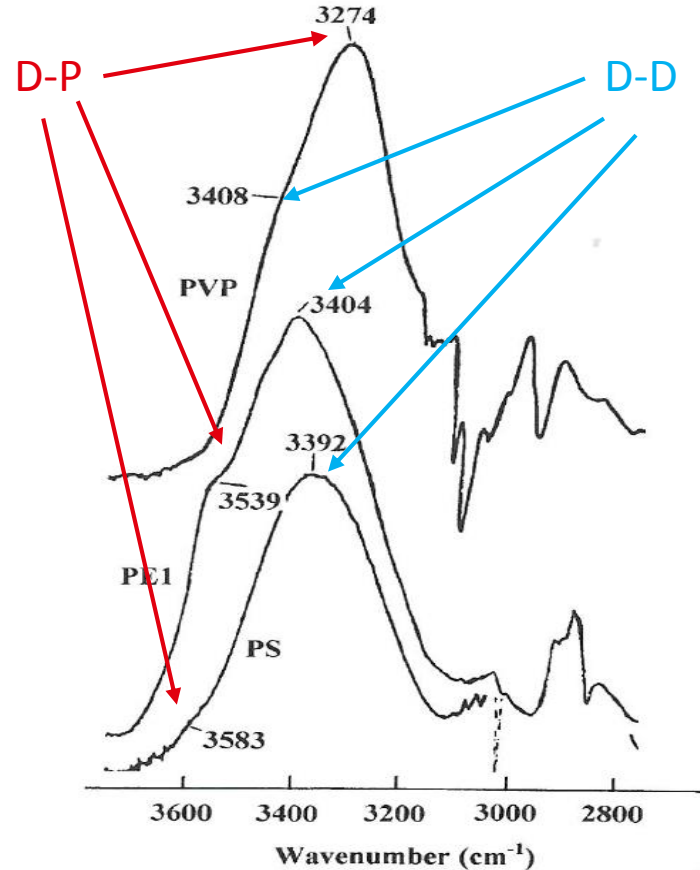
- $-N= > -CON= > -CN > -COO- > -\pi > -Cl$

D-D interaction strength

- reverse of above (small range)

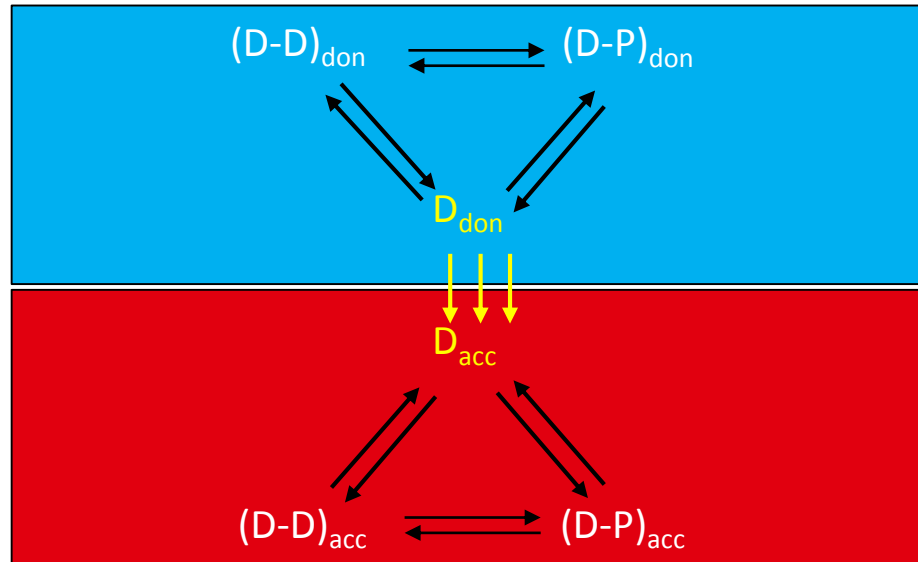
D-P: D-D ratio

- $-N= > -CON= > -CN > -COO- > -\pi > -Cl$



| Dye self-association and dye-polymer interaction

- IR shows complexity, D-D and D-P
- studies done at room temperature
- equilibria will be temperature dependent



| Conclusions

Dye-polymer interactions in the donor layer

- dye transfer controlled by dye-polymer blend Tg
- higher Tg from stronger dye-polymer interactions and similar solubility parameters

Dye-polymer interactions in the acceptor layer

- solution-diffusion model
- higher dye transfer from lower Tg, similar dye-solubility parameters and stronger specific dye-polymer interactions

Dye-dye interactions...

- increase with the number of dye –OH groups
- decrease with the strength of polymer electron donor

| Acknowledgements

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Thank you!